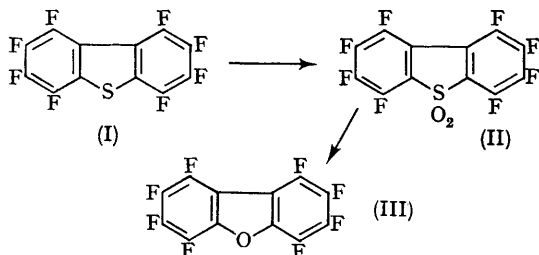


An SO Elimination leading to Octafluorodibenzofuran

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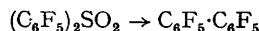
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We recently reported¹ the first synthesis of a perfluorinated compound in the dibenzo-series, octafluorodibenzothiophen (I), and we have

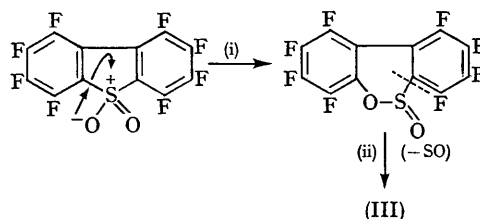


subsequently found that a dioxide (II) can be formed by the reaction of (I) with 85% hydrogen peroxide in trifluoroacetic anhydride, while other methods gave little or incomplete reaction. Current interest in the elimination of sulphur monoxide from sulphones² led us to examine the mass spectrum of the dioxide (II) and a large peak at 312 mass units, corresponding to P - SO was observed. Subsequently, we pyrolysed the dioxide (II) and found that the new heterocycle, octafluorodibenzofuran, m.p. 100°, could be isolated in 72% yield. The pyrolysis temperature and conditions were critical; general decomposition occurred on pyrolysis in a sealed tube but, using a flow system, products were obtained which became more simple as the temperature was raised. At 810° octafluorodibenzofuran was obtained in high

yield together with small amounts of octafluorodibenzothiophen and a minor unknown impurity. In contrast to these results, pyrolysis of decafluorodiphenyl sulphone (IV), under various conditions, gave only sulphur dioxide elimination, leading to decafluorobiphenyl.



The mechanism of the SO elimination possibly involves an intramolecular rearrangement (internal nucleophilic displacement) to an unstable sulphinic ester, followed by SO loss.



Elimination of SO or SO₂ from a compound would then depend on the relative rates of rearrangement (i) to C-S bond cleavage (ii). These possibilities are being investigated.

Octafluorodibenzofuran undergoes nucleophilic substitution by a variety of nucleophiles but the orientation, while of considerable interest, cannot yet be unambiguously assigned.

(Received, April 28th, 1967; Com. 405.)

¹ R. D. Chambers and J. Cunningham, *Chem. Comm.*, 1966, 469.

² Results communicated by J. F. W. McOmie and by M. P. Cava at the Symposium on "Aromaticity", Sheffield, 1966.